

Materials Aspects and Application of Superconductivity

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Brief introduction to superconductivity

Electrical Resistance (Resistivity)

🍏 Ohm's law: $V=RI$

♠ V : voltage, I : current, R : resistance

♠ R depends on amount of material.

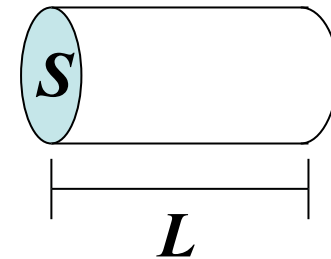
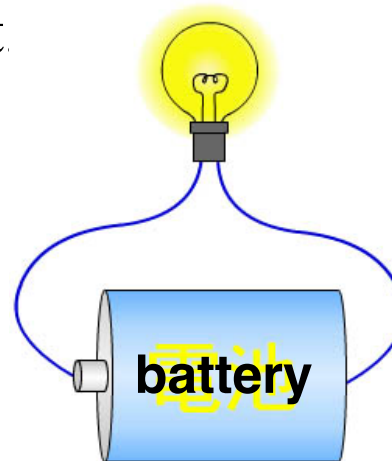
🍏 $R=\rho L/S$

♠ ρ : resistivity, L : length, S : cross-section

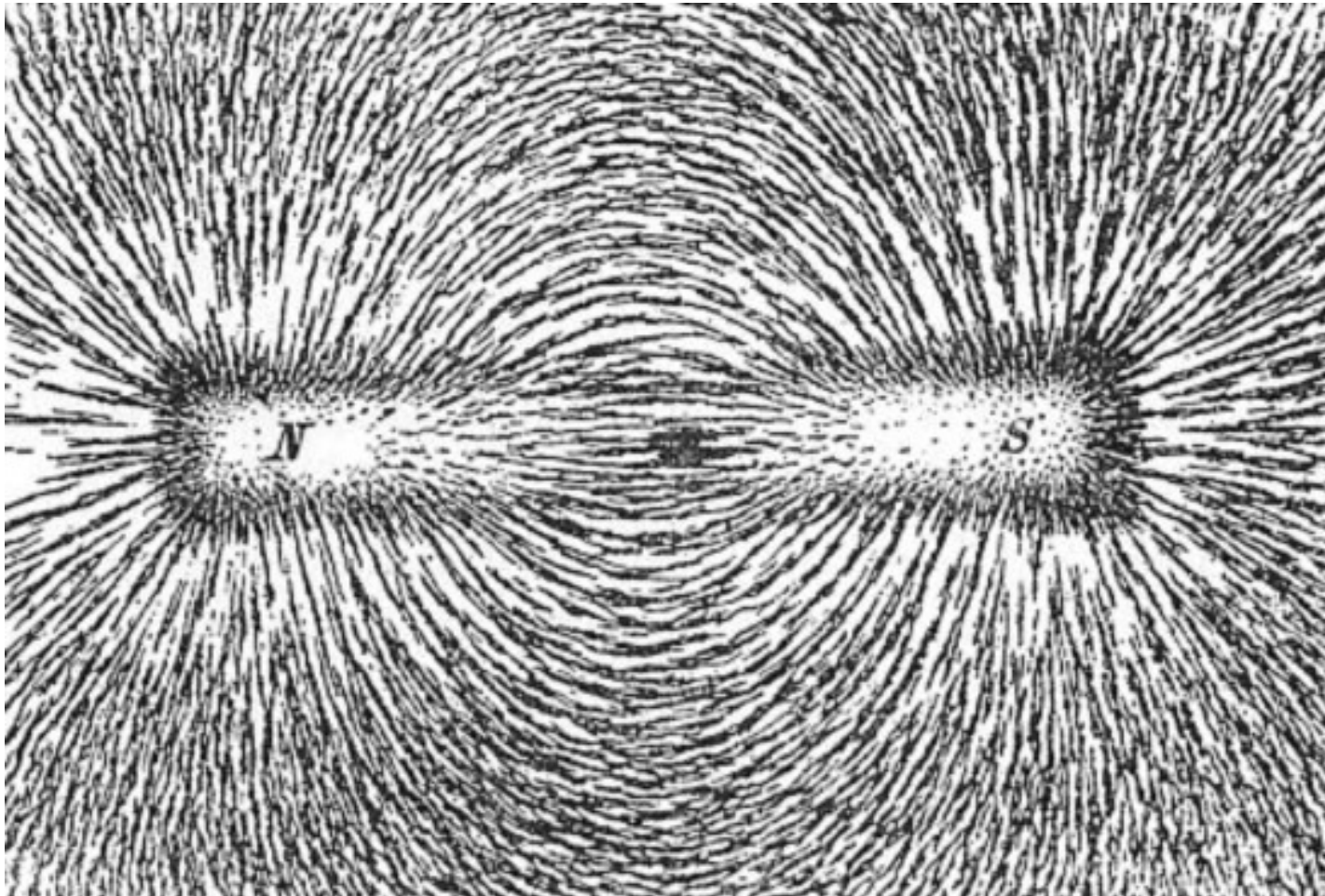
♠ ρ is material constant

🍏 Joule heat (energy)

♠ $W=I^2R$



Magnetic force lines



What is superconductivity?

- 🍏 In some materials, electrical resistance is absolutely zero below a certain temperature. This phenomenon is called "**superconductivity**" and this characteristic temperature is called "**critical temperature**" or "superconductivity transition temperature" (T_c).
- 🍏 T_c is very low in our normal sense (its present world record at ambient pressure is 133 K (-140°C)), however, if T_c is higher than 77 K (-196°C), we can use cheap liquid nitrogen instead of too expensive liquid helium (4.2 K) for cooling.

1911: discovered by H. K. Onnes

(Leiden Univ., Netherland)

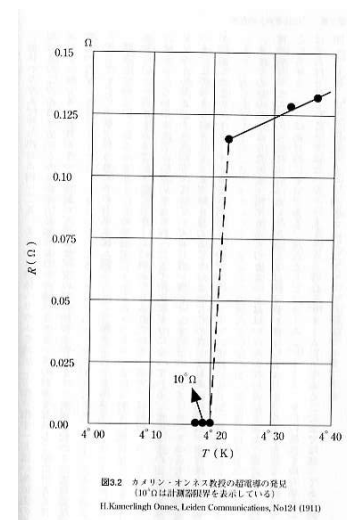
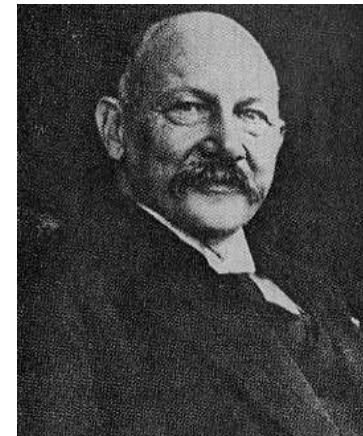
🍏 Resistivity of metals

♠ decreases as temperature (T) lowers.

🍏 At absolute zero (0 K, -273°C), does it go down to zero or diverge to infinity?

♠ Using highly pure metal (mercury; Hg) and liquid helium (liq. He), Onnes measured temperature-dependence of resistivity, and then, discovered "Superconductivity".

♠ Hg is liquid at room-temperature, so Hg is easily vaporized. By re-condensation, highly pure Hg can be obtained.



Characterics

(on the standpoint of Physics)

🍏 Characteristics of superconductivity is not only "zero resistance" but also ...

♠ zero Joule loss: energy saving

🍏 Perfect Diamagnetism: type I superconductor

♠ exclude magnetic flux

♠ In type II, this is "imperfect".

🍏 Quantum Vortex

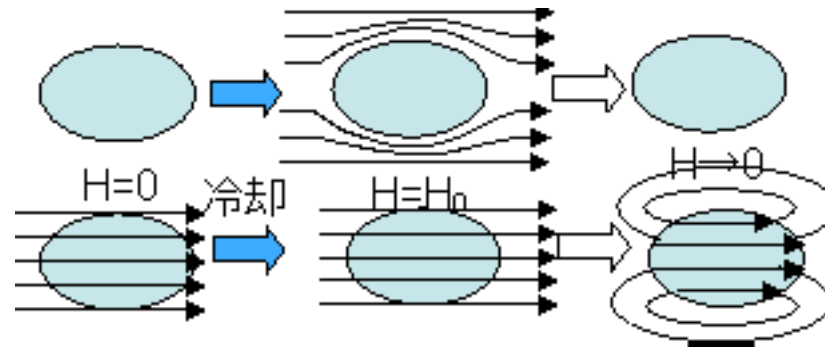
🍏 Josephson Effect

Type I Superconductor \neq Perfect conductor



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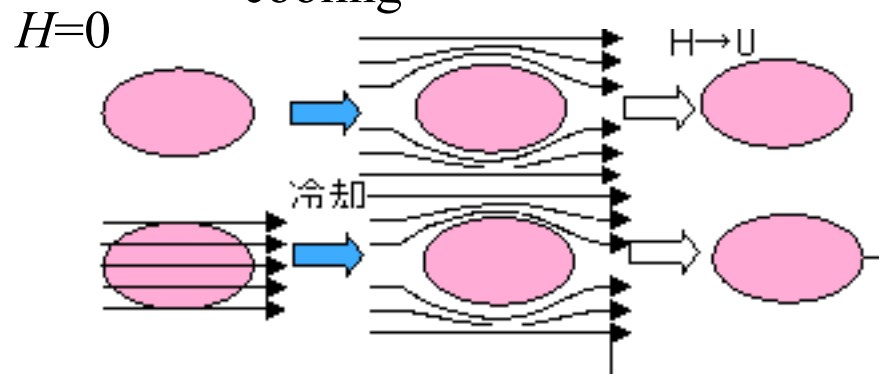
room-temperature



perfect conductor

$$\frac{\partial \mathbf{B}}{\partial t} = 0$$

cooling



superconductor

$$\frac{\partial \mathbf{B}}{\partial t} = 0$$

$$\mathbf{B} = 0$$

by London

normal state

superconducting state

Superconductor \neq Perfect conductor



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🍏 Maxwell's equations

$$\nabla \cdot \mathbf{D} = \rho$$

$$\nabla \times \mathbf{E} = -\frac{\partial \mathbf{B}}{\partial t}$$

$$\nabla \cdot \mathbf{B} = 0$$

$$\nabla \times \mathbf{H} = -\frac{\partial \mathbf{D}}{\partial t} + \mathbf{J}$$

♠ $\mathbf{D} = \epsilon_0 \mathbf{E} + \mathbf{P} = \epsilon_r \epsilon_0 \mathbf{E} = \epsilon \mathbf{E}, \mathbf{B} = \mu_0 \mathbf{H} + \mathbf{M} = \mu_r \mu_0 \mathbf{H} = \mu \mathbf{H}$

🍏 We can conclude that "*superconducting state*" and "*normal state*" are different phases from each other.

Why superconductivity: Application



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Zero-resistance

- ♠ very strong electromagnet (NMR, MRI)
- ♠ energy storage (SMES; superconducting magnetic energy storage)

Diamagnetism

- ♠ magnetic shielding

Quantum Vortex

- ♠ highly sensitive magnetic sensor (SQUID *etc.*)

Josephson effect

- ♠ highly sensitive magnetic sensor (SQUID *etc.*)

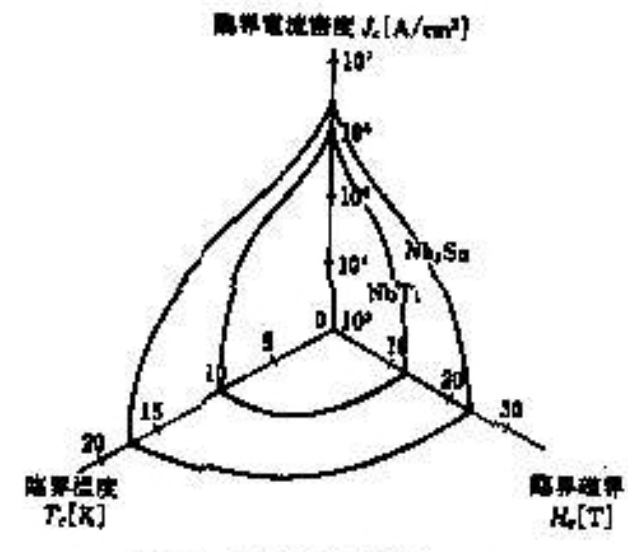
3 critical parameters

🍏 Critical temperature (T_c)

🍏 Critical field (H_{c1} , H_{c2})

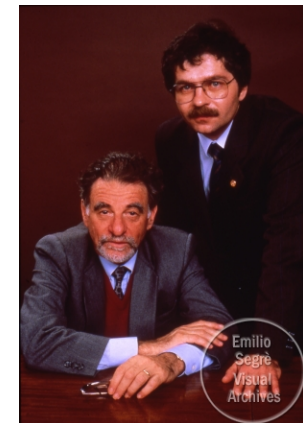
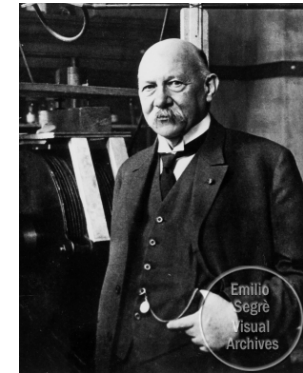
🍏 Critical current (J_c)

♠ the most important parameter for practical application

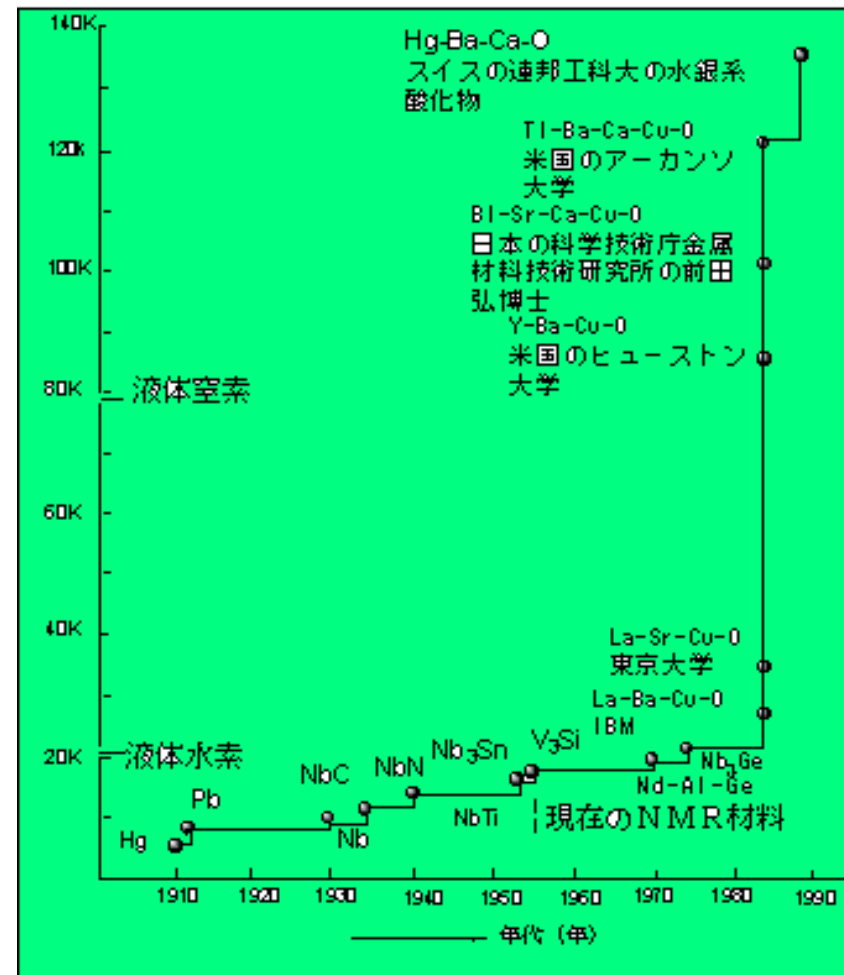


Brief History of Superconductivity

- 🍏 1908: Liquid Helium
- 🍏 1911: discovery of Superconductivity (Hg, Sn, Pb)
- 🍏 1933: Meissner effect
- 🍏 1935: London equation
- 🍏 1935-37: Type II Superconductor
- 🍏 1954: Nb₃Sn (18.5 K)
- 🍏 1960: GL theory
- 🍏 1961: Quantum Vortex
- 🍏 1961: NbTi (9.8 K)
- 🍏 1957: BCS theory
- 🍏 1962: Josephson effect
- 🍏 1974: Nb₃Ge (23.2 K)
- 🍏 1986: discovery of High-Temperature Superconductivity



Increasing T_c : HTSC



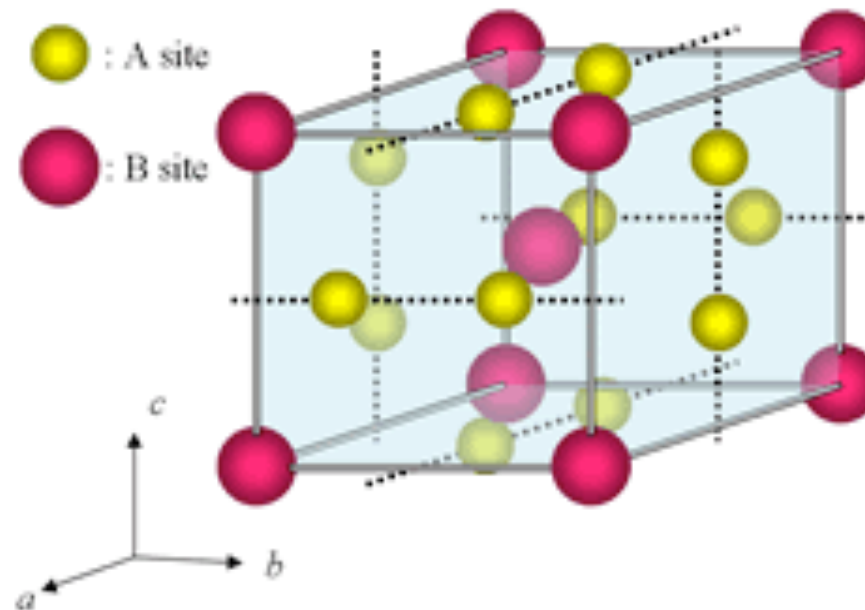
Materials Aspects

Other (Compound) superconductors

- 🍏 Alloys: NbTi *etc.*
- 🍏 Intermetallic compounds: Nb₃Sn, Nb₃Al *etc.*
- 🍏 Organic compounds: (TMTSF)₂PF₆ *etc.*
- 🍏 Sulfides: PbMo₆S₈ *etc.*
- 🍏 Oxides (Cuprates: high-temperature superconductors)
 - ♠ SrTiO₃, Ba(Pb,Bi)O₃ *etc.*
 - ♠ (La,Ba)Cu₂O₄
 - ♠ YBa₂Cu₃O₇
 - ♠ Bi₂Sr₂Ca_{n-1}Cu_nO_{2n+4}, Tl₂Ba₂Ca_{n-1}Cu_nO_{2n+4} *etc.*
 - ♠ Fe-based Oxides (Arsenides): new comers

Conventional Superconductors

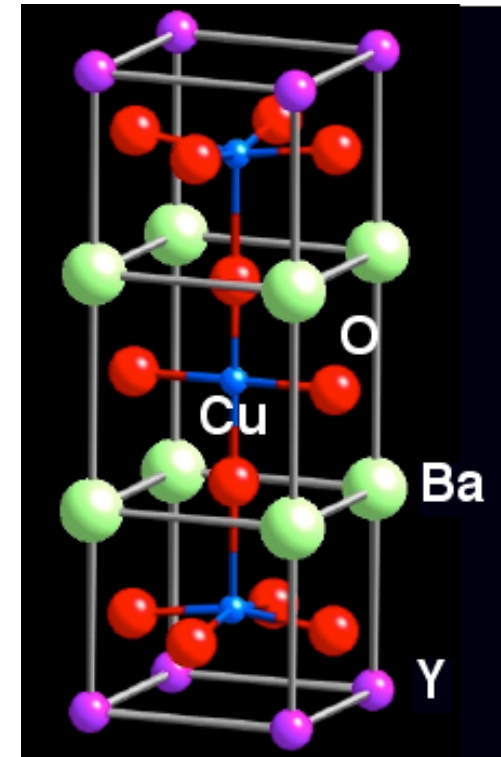
- 🍏 At present, superconductors used for practical application are NbTi (bcc) and Nb₃Ti only.



Nb₃Sn: A15 structure

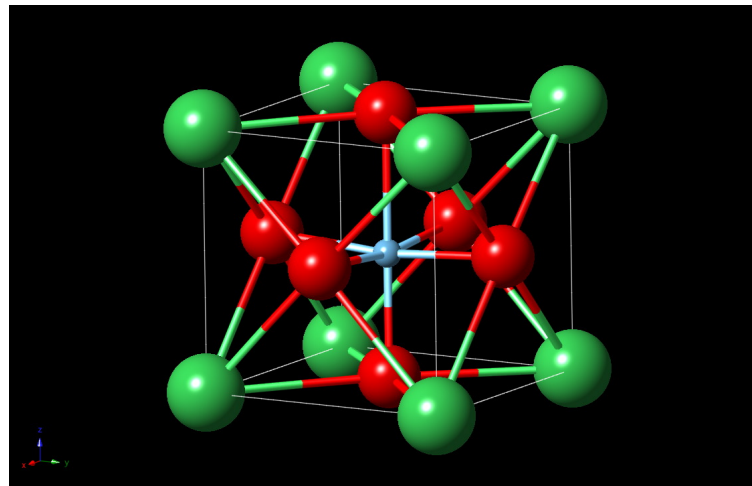
$\text{YBa}_2\text{Cu}_3\text{O}_z$

- 🍏 The first material having T_c higher than 77 K ($T_c \approx 90$ K for $z \approx 7$; quite sensitive on z value).
- 🍏 The structure is oxygen-deficient triple perovskite.



Perovskite structure

- 🍏 Ionic crystal with chemical formula of ABO_3 .
- 🍏 Many functional oxides with 3d transition metals.
- 🍏 The most famous one is (probably) BaTiO_3 in which TiO_6 octahedron is slightly distorted.



Examples of Application

Labo. application



High-field magnet



SQUID



NMR

Medical application (MRI)

Magnetic Resonance Inspection



Transportation (MAGLEV)

Magnetically Levitation Train



Apr. 21, 2015: World record: 603 km/h

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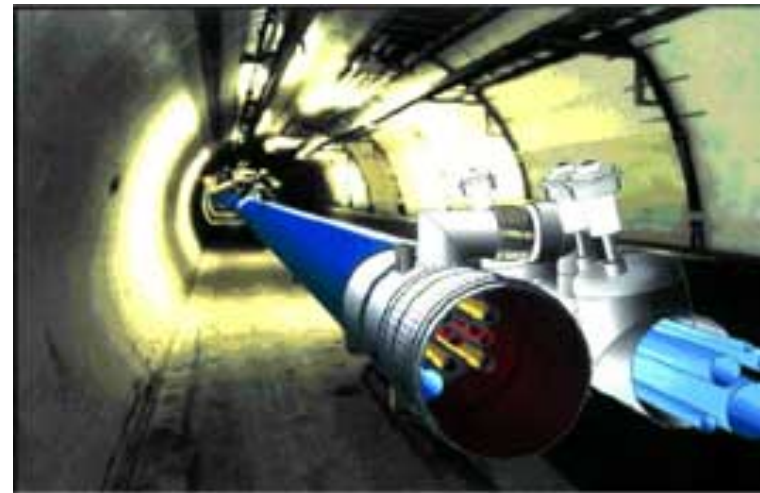
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Large scale application

LHC: Large hadron collider (CERN)

 Nobel prize in 2013

 Japanese suppliers contribute a lot.



Superconducting power cable

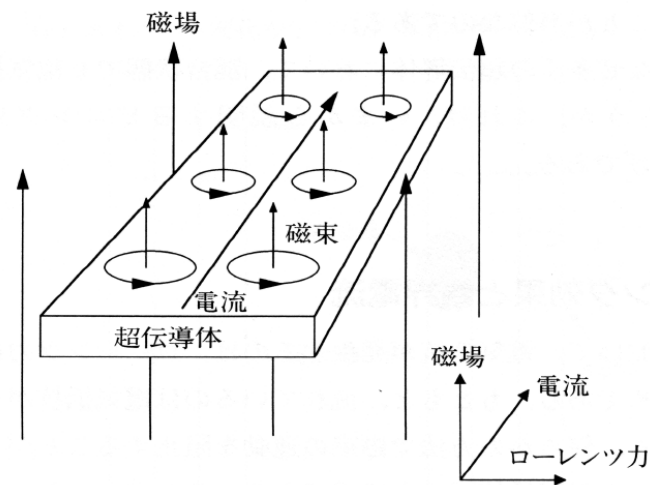
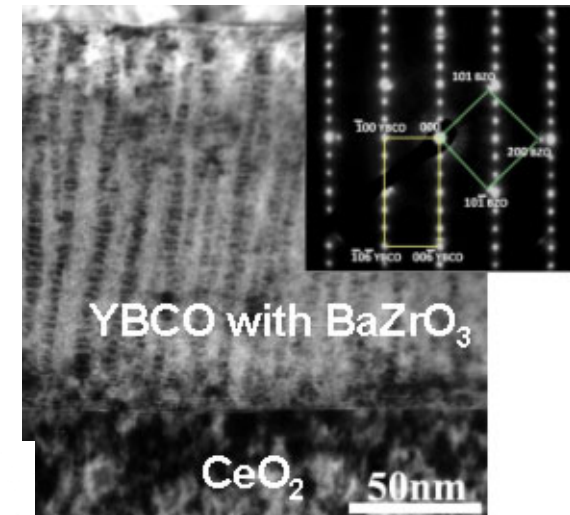
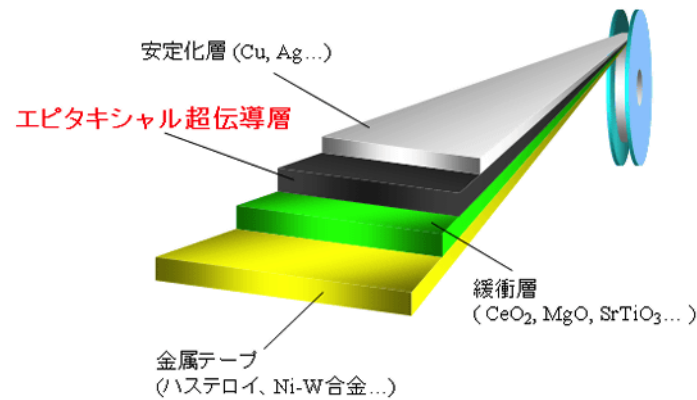


1st generation wire: Ag-sheathed $\text{Bi}_2\text{Sr}_2\text{Ca}_2\text{Cu}_3\text{O}_{10}$: powder sintering

2nd generation wire (epitaxial thin film)



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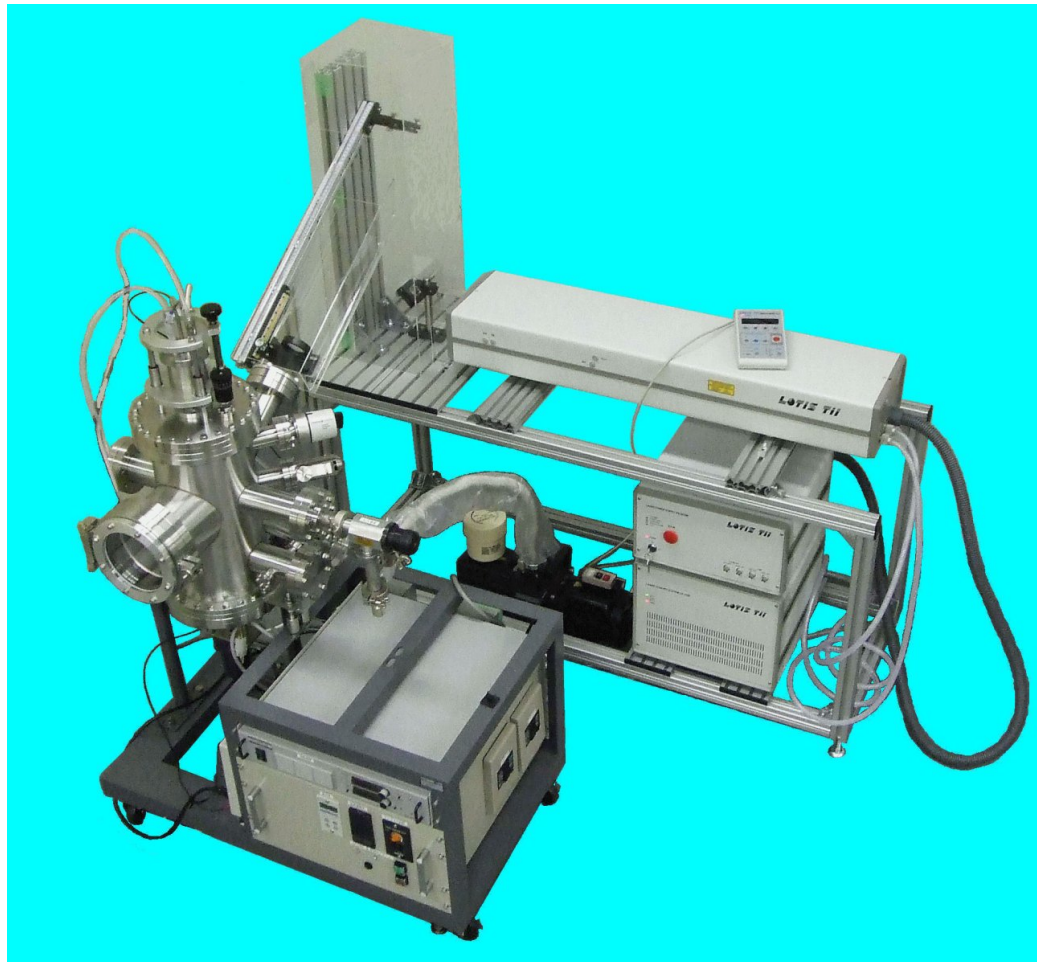


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PLD apparatus

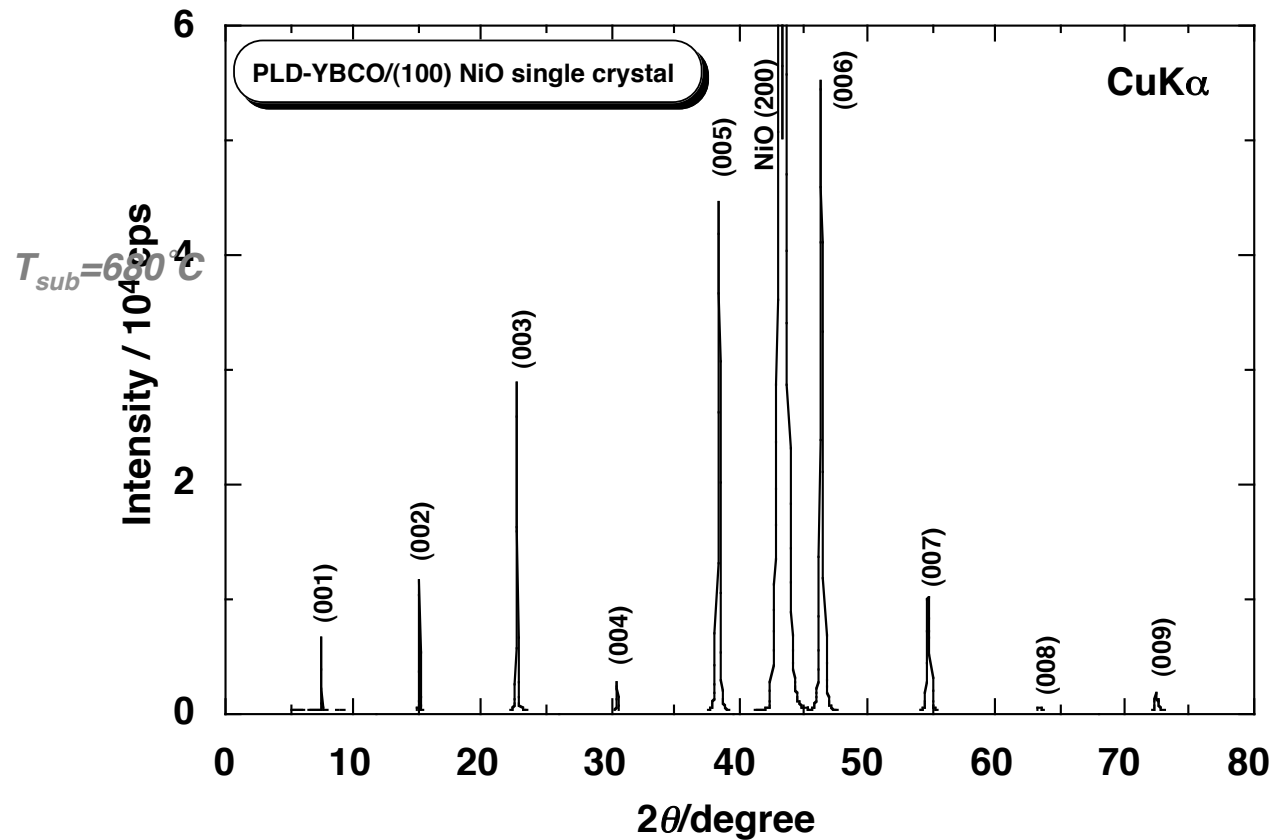


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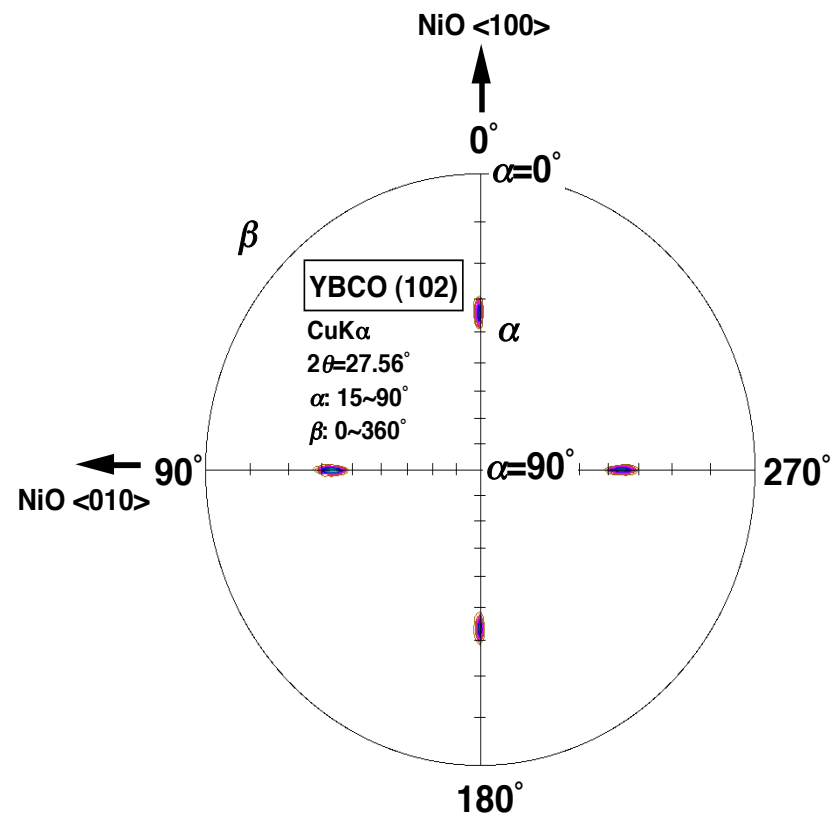
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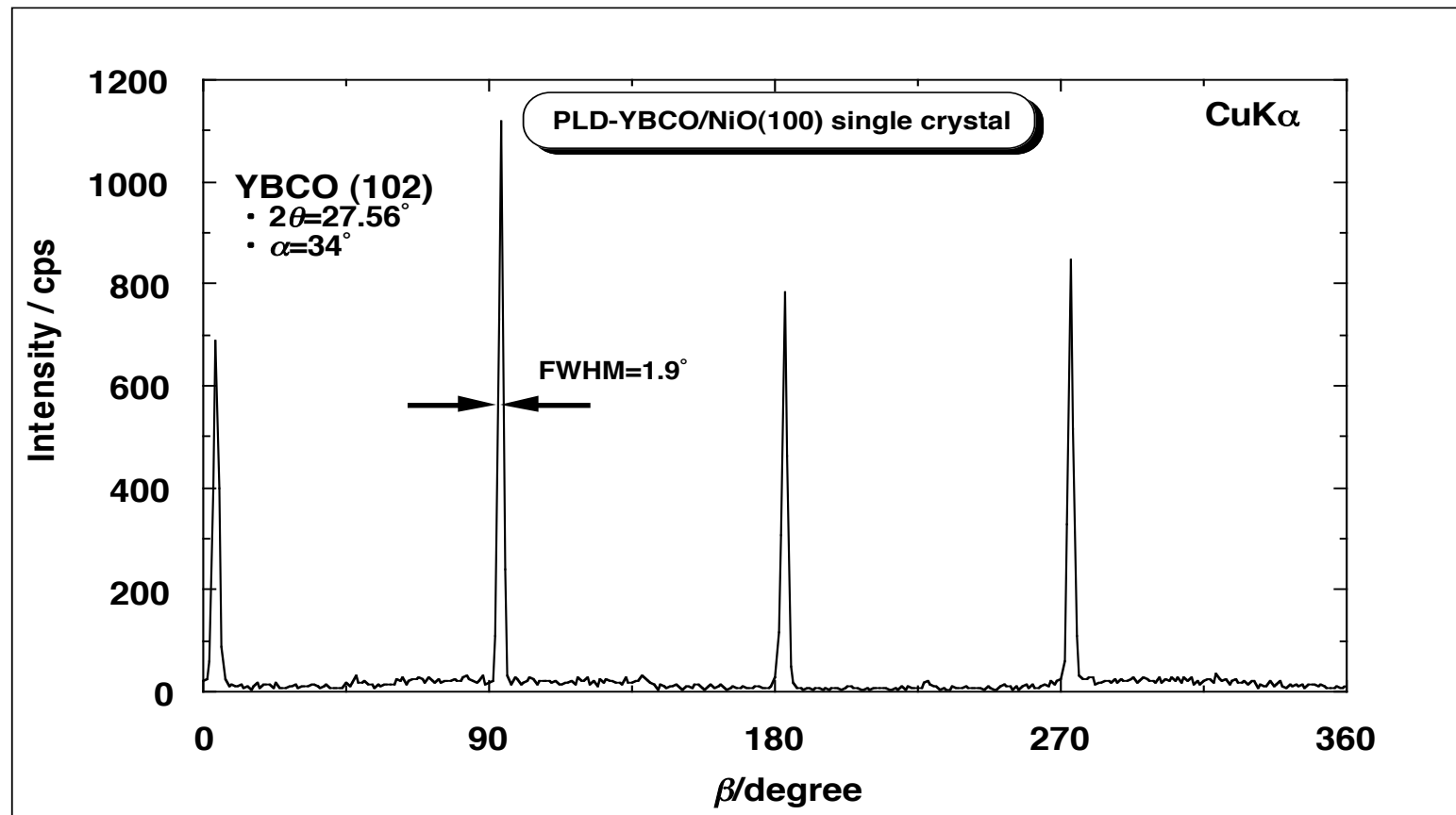
XRD profile of YBCO film



Pole figure



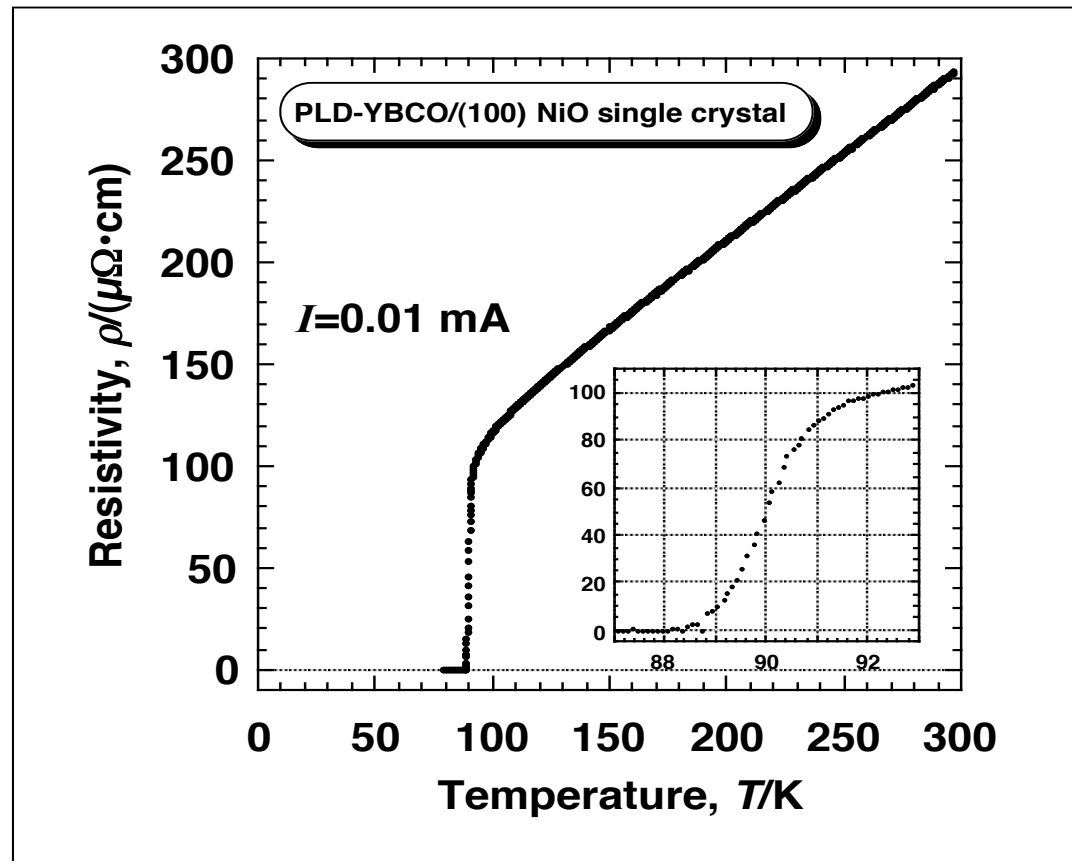
β -scan profile



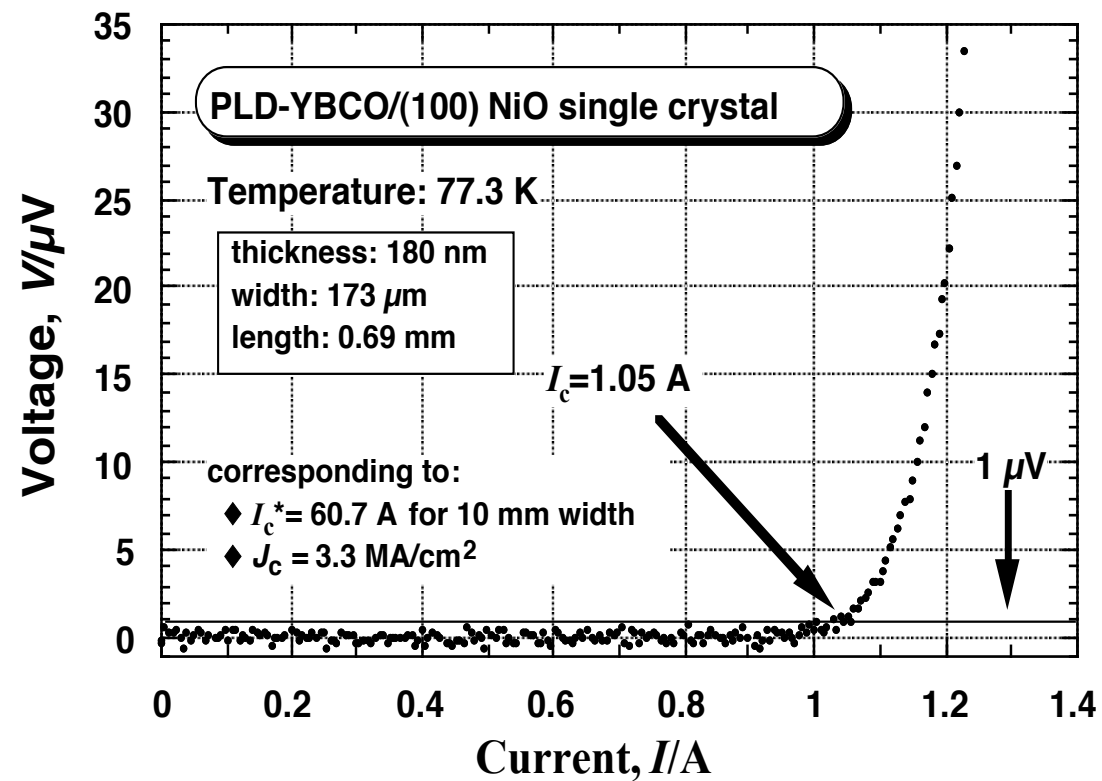
Temperature dependence of resistivity



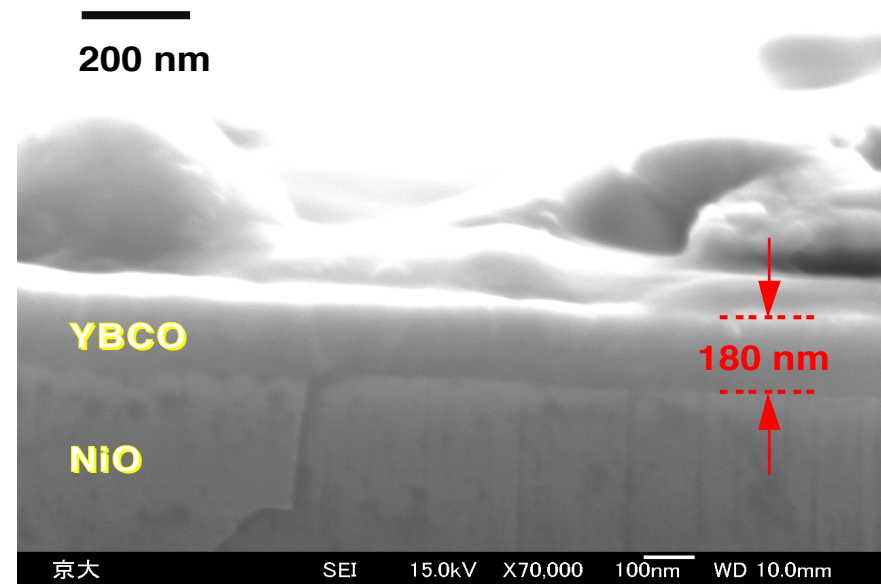
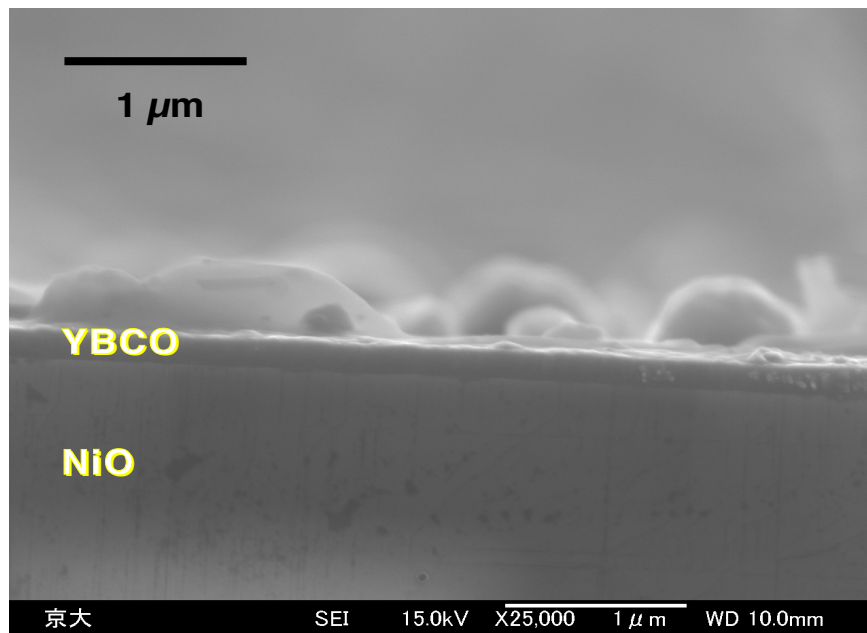
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I-V characteristics at 77 K

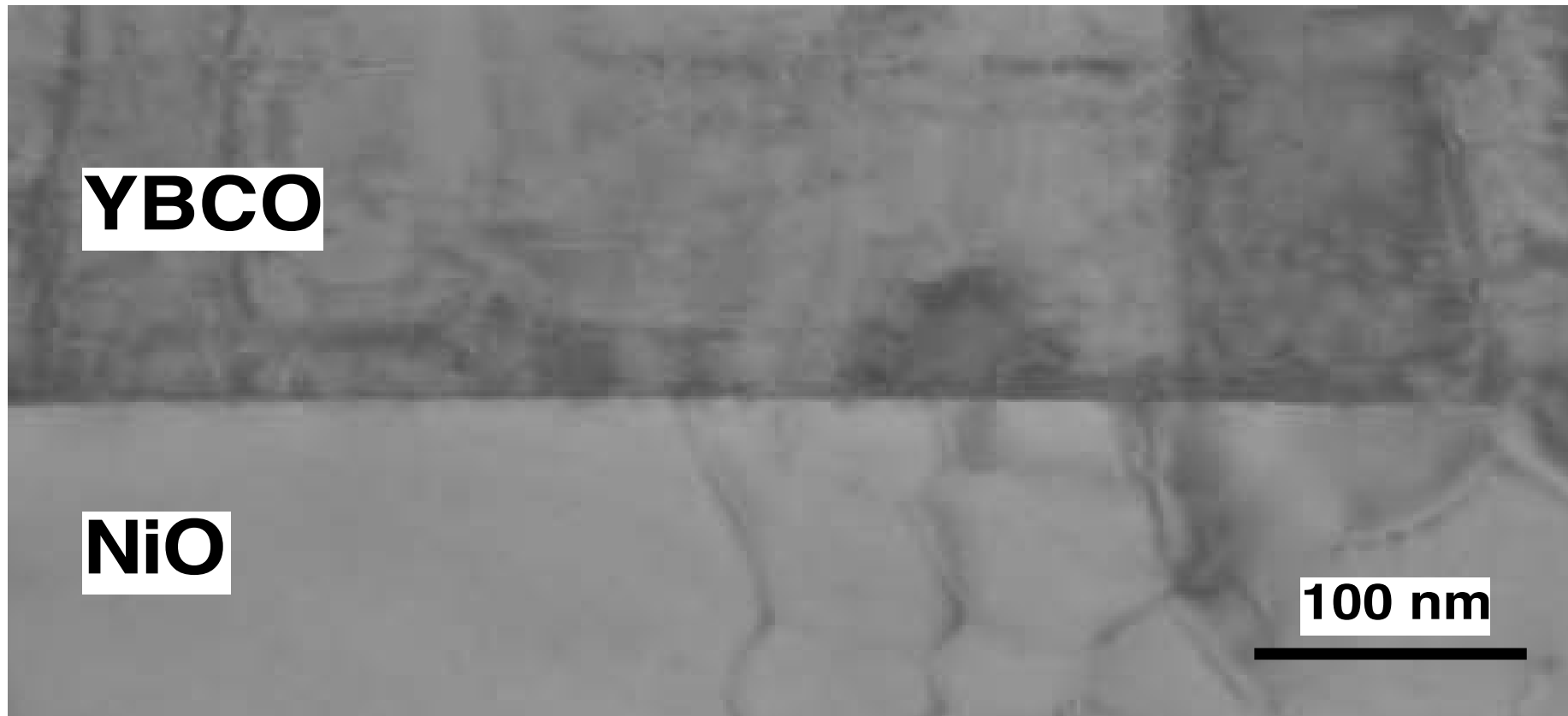


Cross sectional HRSEM images of fractured surface for PLD-YBCO/NiO structure



(taken by Kyoto Univ.)

TEM image of interface structure of PLD-YBCO/(100) NiO single crystal



Summary

Superconductivity

 hopeful

 interesting

 cute

That's all !!

Thank you for your attention !!

Submit your report at the beginning of
next lecture.